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CHEMISTRY IX NOTES ELECTROCHEMISTRY – MULTIPLE CHOICE QUESTIONS

9. The unit of electro chemical equivalent is
a. Ampere b. Coulomb
c. G/coulomb d. V kg/coulomb

10. In dry cell electrolyte is the moist paste of
a. $\text{NH}_4\text{Cl} + \text{MnO}_2$ b. $\text{MnO}_2 + \text{ZnCl}_2$
c. $\text{NH}_4\text{Cl} + \text{ZnCl}_2$ d. $\text{NH}_4\text{Cl} + \text{ZnSO}_4$

11. A weak electrolyte is:
a. Sulphuric b. Nitric acid
c. A caustic alkali d. Citric acid

12. The electrolysis of fused sodium chloride is carried out in:
a. Galvanic cell b. Voltaic cell
c. Down's cell d. Dry cell

13. The electrodes used in the electrolytic cell to carry out electrolysis of water are made of:
a. V Platinum b. Carbon
c. Graphite d. Copper

14. During electrolysis of which hydrogen is released at the cathode and oxygen is released at the anode?
a. V Water b. NaCl
c. HgO d. H_2O_2

15. One Faraday has
a. V 96,500 C b. 94,500 C
c. 93,9500 C d. 92,500 C

16. The deposition of a metal on another metal is called:
a. Electrolysis b. Oxidation
c. Reduction d. V Electroplating

17. In electroplating, the thing to be electroplated is made the:
a. V Cathode b. Electrolytic cell
c. Anode d. Electrolyte

18. In which cells, oxidation loss of electrons occurs at anode and reduction gain of electrons occurs at cathode?
a. V Electrochemical b. Electrolytic
c. Primary d. Secondary

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